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Terms in this set (32) Which one of the following reactions is NOT balanced?
 $2 \text{CO} + \text{O}_2 \rightarrow 2 \text{CO}_2$
 $2 \text{SO}_2 + \text{O}_2 \rightarrow 2 \text{SO}_3$
 $\text{SF}_4 + 3 \text{H}_2\text{O} \rightarrow \text{H}_2\text{SO}_3 + 4 \text{HF}$
 $2 \text{KNO}_3 + 10 \text{K} \rightarrow 5 \text{K}_2\text{O} + \text{N}_2$

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ANSWER: Correct Chapter 9 Reading
Question 2 Part A Iron reacts with oxygen in
the air to produce rust (Fe_2O_3) according to
the reaction shown below. $4 \text{Fe} + 3 \text{O}_2 \rightarrow 2$
 Fe_2O_3 How many moles of rust can be
formed if 1.8 mols of iron are present? Hint
1. Begin with your given number of moles.

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CHAPTER 8 REVIEW Chemical Equations

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Chapter 7
and Reactions SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1. List four metals that will not replace hydrogen in an acid. Choose from Cu, Ag, Au, Pt, Sb, Bi, and Hg.

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Chapter 7
most to fewest valence electrons of the following elements: Kr, Ga, Br, As, K. -
Answer: Kr (8), Br (7), As (5), Ga (3), K (1).
Valence electrons are equal to the atom's main group number. Link: <https://www.coursehero.com/file/20333255/Mastering-Chemistry-Chapter-2/>.

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Start studying Mastering Chemistry Chapter 3. Learn vocabulary, terms and more with flashcards, games and other study tools. On the periodic table, one of the columns within the main group elements; a family or group of elements exhibits similar chemical properties. (3.2). <https://quizlet.com/23732>

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