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**of elements in 30 minutes |**

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Table Explained:

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Notes

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law of conservation of mass.

law of definite proportions.

law of multiple proportions.

atom. States that matter can

be neither destroyed nor

created during... The fact

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that a chemical compounds  
contains the same elements...  
if two or more different  
compounds are composed of  
the same tw... .

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Atoms. • Atoms are building blocks of all matter. •

According to modern atomic theory, an atom is the smallest particle of an element which takes part in chemical reaction. • Atoms

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are very small and which can't be seen even through very powerful microscope. • Atomic radius is measured in nanometres.

Notes of Ch 3 Atoms and  
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MIXED REVIEW. SHORT ANSWER

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Answer the following questions in the space provided. 3 points each, 9 points total. 1. The element boron, B, has an atomic mass of 10.81 amu ...

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Mixed Review Answers

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Chapter 11 Modern Atomic

Theory 1 GPS: SC3 Students

will use the modern atomic

theory to explain the

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characteristics of 2 atoms.

0 a. Discriminate between the relative size, charge, and position of protons, 1 neutrons, and electrons in the atom. 2 b. Use the orbital configuration of neutral atoms to explain its

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effect ...

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CHAPTER 3 REVIEW Atoms: The  
Building Blocks of Matter  
SECTION 2 SHORT ANSWER

*Page 27/54*

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Answer the following questions in the space provided.

1. In cathode-ray tubes, the cathode ray is emitted from the negative electrode, which is called the cathode .

2. The smallest unit of an element

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that can exist either alone  
or in molecules containing  
the

3 Atoms: The Building Blocks  
of Matter

Advantages of pH Calculation  
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Electrochemistry Notes.  
Studying from Class 12  
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helps you to calculate pH  
easily because: 1) It is  
simple and easier to set up.  
2) It does not require  
constant attention. 3)

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Equilibrium is quickly attained, hence it is not so time-consuming. 4) A small quantity of solution is needed.

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is now 17 years old and ready to become a student at the University of Ingolstadt in Ingolstadt, Germany (near Munich), but an outbreak of scarlet fever at home delays his departure. His mother and "cousin" both fight the

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disease; Caroline Beaufort  
Frankenstein dies, and  
Elizabeth recovers.

Chapter 3

Taken from the book Modern  
Chemistry by Holt, Rinehart,  
and Winston on Chapters 4

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and 5, which deals with electrons and the periodic table. Includes the chapter vocabulary and a few other useful things.

Modern Chemistry (Holt,  
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Inhibitors in HIV Drugs,  
Enzyme-Substrate Binding,  
Vision and Light Induced

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