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and Related gas laws question_answer At what temperature, the sample of neon
gas would be heated to double of its pressure, if the initial volume of gas is/are
reduced to 15% at $\{75\}^{\{0\}}\text{C}$ [Kerala CET 2005]

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question_answer Which of the following statement is false [BHU 1994] . A) The
product of pressure and volume of fixed amount of a gas is independent of
temperature . B) Molecules of different gases have the same K.E. at a given
temperature

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of moles of gas present in the container of volume 10 lit at 300 K. If the manometer containing glycerin shows 5m difference in level as shown in diagram.

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Matter Total Question - 50 question_answer1) _____ is water in its gaseous form. ...
The conversion of solid into gas without changing into liquid is known as _____ A)
Heating done clear. B) Evaporation done clear. C) Melting ...

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ACS Gas Questions - Gas Safe Training

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Question 11. How Natural Gas Is Generally Traded? Answer : Gas trade is based on \$/MMBTU (Million British Thermal Units) What is the composition of natural Gas: Composition of Natural gas varies widely from source to source. Natural gas is a combustible mixture of hydrocarbon gases.

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(i) It is an inflammable gas (ii) It is the lightest gas. (iii) It is not easily liquefied. (iv) It is a strong oxidizing agent. (e) For the reaction $\text{PbO} + \text{H}_2 \rightarrow \text{Pb} + \text{H}_2\text{O}$, which of the following statements is wrong? i) H_2 is the reducing agent. (ii) PbO is the oxidizing agent. (iii) PbO is oxidized to Pb (iv) H_2 is oxidized to H_2O .

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Study the following figure and answer questions. a) After heating Calcium carbonate, which gas is formed in a test tube? b) When we pass this gas through limewater what change, did you observe? c) Write down the chemical reaction showing the product formation after heating the Calcium carbonate.

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